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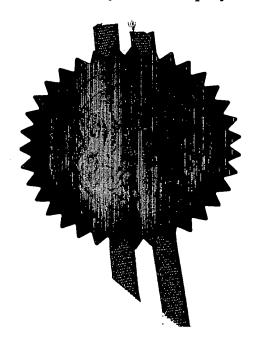
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Dated 20 October 2004

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11SEP03 E83665 P01/7700 0.00-0321286.7

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Patent application number (The Patent Office will fill in this part) SEP 2003

0321286.7

Full name, address and postcode of the or of each applicant (underline all surnames)

08412010001

Patents ADP number (If you know It)

If the applicant is a corporate body, give the country/state of its incorporation

Safeglass (Europe) Limited Whitworth Building Scottish Enterprise Technology Park East Kilbride GLASGOW G75 0QD

UK

Title of the invention"

improved container

5. Name of your agent (If you have one)

*Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)

Kennedys Patent Agency Limited Floor 5, Queens House 29 St Vincent Place Glasgow **G1 2DT**

Patents ADP number (If you know it)

0805 824 0002

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Description

Claim (c)

Abstract

Drawing (4)

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Priority documents

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A. . . Request for substantive examination Patents Form 10/77)

> 1 Any other documents (please-specify)

I/We request the grant of apparent on the basis of this application.

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11 September 2003

12. Name and daytime telephone number of person to contact in the United Kingdom Karen Veitch

Tel: 0141 226 6826

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1

The present invention relates to a container for
foodstuffs and beverages which has improved safety
characteristics when compared to conventional containers.

Improved Container

7 Traditionally, the containers in which alcoholic drinks,

8 carbonated soft drinks and oxygen sensitive juices are

9 sold are manufactured from glass. Glass bottles are well

10 received by consumers as they impart the impression of a

11 high quality product and have "chink factor".

12 Nevertheless, the use of bottles is inherently dangerous,

13 as glass is easily broken. It will be appreciated that

14 this is a particular problem in bars, pubs and

15 nightclubs, where accidental breakage of glass bottles,

16 is a potential health risk.

18 Glass bottles are also disliked as they can be used

19 deliberately, as weapons, to inflict damage on other

20 persons. In fact, safety regulators have actively

21 encouraged drinks manufacturers, as well as

22 establishments which serve drinks and alcohol, to use

23 bottles and glasses made from non-dangerous materials, in

1 order to reduce the number of serious injuries caused by

2 glass and bottle attacks.

3

- 4 In recent years there has been a move towards providing
- 5 bottles manufactured from materials which are not as
- 6 dangerous as glass. It is estimated that packaged beer
- 7 production world wide in 1996 was 106.6 billion litres,
- 8 requiring 186.2 billion bottles and 73.7 billion cans.
- 9 The beer bottle marker was forecasted to grow at an
- 10 annual rate of three percent through 2001 to 216 billion
- 11 units. Most bottle production makes use of glass with
- 12 only 0.1 billion plastic bottles being utilised in 1996.
- 13 However due to the push towards increased safety it is
- 14 estimated that the demand for plastic bottles is forecast
- 15 to reach 2.5 billion by 2006.

16

- 17 The focus on safer bottlesgis particularly important with
- 18 respect to alcoholic drinks, which are consumed in bars
- 19 and nightclubs. PET, poly(ethylene terephthalate), a
- 20 plastic which can be readily manufactured into bottles,
- 21 and which does not break as readily as glass, has already
- 22 been used for this purpose.

23

- 24 However the use of PET poses its own problems to the
- 25 industry. PET is a relatively expensive material and not
- 26 cheap to processwhich makes it a less popular option for
- 27 drinks manufacturers. In addition, there is a general
- 28 consensus that plastic bottles are not as well received
- 29 by the public as they feel cheaper and do not have the
- 30 same high quality feel as glass.

3

- 1 It is an object of the present invention to overcome the
- 2 problems that are described above with reference to glass
- 3 and existing plastic bottles.

4

- 5 According to a first aspect of the present invention,
- 6 there is provided a container manufactured from a
- 7 material that shatters when broken into fragments which
- 8 do not cut, puncture or otherwise damage human skin or
- 9 tissue.

10

- 11 The container may be a bottle, glass, tumbler, or the
- 12 like.

13

- 14 Preferably the material is comprised of an amorphous
- 15 thermoplastic polymer and one or more low molecular
- 16 weight resins.

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18 Preferably the amorphous thermoplastic polymer is chosen

32. Buch 36.

Land to the world with the configuration of the

19 from the group consisting of: " "

20

- 21 polystyrene (PS)
- 22 styrene-acrylonitrile co-polymer (SAN)
- 23 linear polyesters and co-polyesters
- 24 polycarbonate (PC)

25

- 26 Preferably the one or more low molecular weight resins
- 27 are hydrocarbon resins:

28

- 29 Preferably the one or more low molecular weight resins
- 30 are aromatic hydrocarbon resins

31

- 32 The one or more low molecular weight resins chosen will
- 33 be completely compatible with the chosen polymer. For

۲.

4

1 example, in the case of polystyrene, the low molecular

- 2 weight resin will typically be C9 aromatic hydrocarbon
- 3 resin.

4

- 5 Preferably the material has a tensile stress limit
- 6 between 11 and 60 Nmm⁻².

7

- 8 Preferably the one or more low molecular weight
- 9 hydrocarbon resins are selected from a group consisting
- 10 of:

11

- 12 Norsolene**
- 13 ⋅ Krystalex**
- 14 * Plastolynma
- 15 Endex™
- 16 Piccotex™
- 17 Piccolastic R
- 18 Sukorez**
- 19 Arkon^{ra}

20

- 21 Most preferably the one or more low molecular weight
- 22 hydrocarbon resins are selected from a group consisting
- 23 of; Norsolene W90TK, Norsolene W100TM, Norsolene W110TM,
- 24 Kristalex F85 TM, Kristalex F100 TM, Kristalex F115 TM,
- 25 Plastolyn 240 [™], Plastolyn 290 [™], Endex 155 [™],
- 26 Piccolastic D125 TM, Sukorez 100 TM, Sukorez 120 TM, Arkon
- 27 P100 TM, Arkon P125 TM, Arkon P140 TM, Piccotex 75 TM,
- 28 Piccotex 100 TM or Piccotex 120 TM.

- 30 Preferably the low molecular weight resin will have a
- 31 M (number average molecular weight) such that it has

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- less than 500 repeating units, and preferably less than
- 2 50 repeating units.

3

- 4 The container may be manufactured from the material using
- 5 injection blow moulding and/or injection stretch blow
- 6 moulding techniques.

7

- 8 Alternatively, the container may be manufactured from the
- 9 material using extrusion blow moulding.

10

- 11 Optionally the material of the container may also
- 12 comprise a oxygen barrier. The material of the container
- , 13 may also comprise oxygen scavengers.

. 14

1

- 15 The barrier included in the material of the container may
- 16 be selected from the group consisting of: acrylonitrile-
- methyl acrylate copolymer, ethylene vinylalcohol (EVOH)
- 18 or nylon MXD6.

19.

- 20 Preferably the barrier is Barex **. Most preferably the
- 21 barrier is Barex™ 210 or Barex™ 218.

22

- 23 In the embodiment where nylon MXD6 is used as a barrier,
- 24 the oxygen scavenger may be X-312. Amosorb 3000, or a
- 25 scavenger of MXD6 with metal catalysed oxygen reduction
- 26 chemistry may also be used.

27

- 28 The barrier may be overmoulded or sprayed onto the
- 29 container or alternatively may be included in the
- 30 material of the container, using co-injection techniques.

- 32 The container may also have an inorganic coating. This
- 33 may be a thin layer of amorphous carbon. The inorganic

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March 1985

1 coating may be applied to the inside surface of the

- 2 container. Typically the inorganic coating will be
- 3 applied in a layer of 100 to 200nm thickness. The layer
- 4 may be applied by spraying.

5

- 6 The container may also have an external organic coating.
- 7 The external organic coating may be PVDC or a two
- 8 component epoxyamine.

9

- 10 The container may be manufactured from multiple layers of
- 11 the material. Two or more layers of the container may be
- 12 combined to act as an improved oxygen barrier.

13

-7:

- ... 14 Optionally the material of the container may also include
- AND UV inhibitors, antioxidants, flow modifiers, colour
- 1/16 pigments and brighteners as known in the art.

17

- 18 Preferably as the amorphous thermoplastic polymer is
 - 19 mixed with the one or more low molecular weight
 - 20 hydrocarbons, the glass transition temperature is
 - 21 elevated. Preferably the material of the container has a
 - 22 glass transition temperature of above 80°C.

23

- 24 Potential uses of the container are not limited. For
- 25 example, the container may be used for beer, carbonated
- 26 soft drinks, oxygen sensitive juices, beverages or milk

- 28 A container having improved safety characteristics is
- 29 manufactured from a material comprised of an amorphous
- 30 thermoplastic polymer and one or more resins. The resins
- 31 are aromatic hydrocarbon resins and are selected from a
- 32 group consisting of Norsolene ™, Krystalex ™, Plastolyn
- 33 TM, Endex TM, Sokorez TM, Arkon TM, Piccolastic and

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- 1 Piccotex™, and in particular Norsolene W90™, Norsolene
- 2 W100[™], Norsolene W110[™], Kristalex F85 [™], Kristalex F100
- 3 [™], Kristalex F115 [™], Plastolyn 240 [™], Plastolyn 290 [™],
- 4 Endex 155 [™], Piccolastic D125 [™], Sukorez 100 [™], Sukorez
- 5 120 TM, Arkon P100 TM, Arkon P125 TM, Arkon P140 TM, Piccotex
- 6 75 TM, Piccotex 100 TM or Piccotex 120 TM. In a particular
- 7 embodiment the one or more low molecular weight resins
- 8 are C9 hydrocarbon resins. The one or more low molecular
- 9 weight resins have an $\overline{M_n}$ (number average molecular
- 10 weight) such that it has less than 500 repeating units.
- II In one particular envisaged embodiment the one or more
- 12 low molecular weight resins have less than 50 repeating
- 13 units. The resin or resins chosen will be selected
- 14 on compatibility with the chosen polymer.
- 16 Low molecular weight in resins is a function of the
- 17 length of the chains in the resin. In this case the
- 18 hydrocarbon resins have a very low molecular weight, too
- 19 low in fact for the resins to be of any use as a
- 20 structural plastics material on their own, and are
- 21 difficult to mould. By mixing low molecular weight
- 22 hydrocarbon resin with polystyrene, the stress limit of
- 23 the polystyrene is reduced giving the material the
- 24 characteristics described in the present Application.
- 26 The amorphous thermoplastic polymer is chosen from the
- 27 group consisting of polystyrene (PS, styrene-
- 28 acrylonitrile co-polymer (SAN), linear polyesters and co-
- 29 polyesters and polycarbonate (PC). These can be mixed,
- 30 blended or polymerised with the one or more low molecular
- 31 weight resins. UV inhibitors, dyes, antioxidants, flow
- 32 modifiers, colour pigments and brighteners can also be
- 33 added to change or adapt the appearance of the container.

1 The container herein described has many characteristics 2 similar to an ordinary glass bottle - i.e. clarity, 3 rigidity and brittleness. However when broken, the 4 bottle shatters into fragments which are harmless and 5 cannot be used to cut or pierce human skin. б 7 The material used to manufacture the container is 8 fundamentally a blend of a rigid and normally brittle 9 amorphous thermoplastic with a glass transition 10 temperature Tg at least 50° C above ambient and one or 11 more compatible low molecular weight resins. A rigid and 12 normally brittle amorphous thermoplastic polymer is 13. blended with one or more low molecular weight resins 14 which have a $\overline{M_{\pi}}$ (number average molecular weight) such 15 that the resin has less than 500 repeating units, is" preferably less than 50 repeating units. The one or more 17 low molecular weight resins have a weight average 18 molecular weight of 6050 or below. The material is, by 19 design, manufactured to break between 11 and 60 Nmm-2. 20 21 The material can be heated and made into the desired 22 shape of the container, i.e. a bottle, glass or tumbler, 23 by any suitable technique known to the art e.g. injection 24 moulding, extrusion blow moulding or pre-form injection . 25 blow moulding techniques. 26 27

28 The container may be manufactured from one or more layers

29 of the material. More than one layer may be used to

30 provide improved oxygen barrier characteristics.

31 Alternatively the container may be coated with an oxygen

32 barrier. Conventional coating technologies can be broadly

33 divided into two categories. The first are those that

1 use vacuum or plasma routes to deposit very thin films of

- 2 materials, such as carbon or silica, onto the surface of
- 3 the article being coated. The second, rely on the
- 4 atomised spraying of liquid organic materials onto the
- 5 external surfaces of the bottle. Ideally all coating
- 6 materials must not interfere with the economics of
- 7 recycling, nor detract from the bottle's appearance, but
- 8 a significant further consideration with thin film
- 9 internal deposits is the need for the materials to be
- 10 approved for food contact.

11

- 12 As the container described herein is manufactured from
- 13 the material at lower processing materials than
- 14 conventional plastics, barriers which are not usually
- 15 suitable for this purpose can be used. For example the
- 16 container can be coated in Barex (acrylonitrile-methyl
- 17 acrylate copolymer), and in particular Barex™ 210 or
- 18 Barex 218, which has high oxygen barrier properties.
- 19 This can be achieved either by overmoulding, spraying or
- 20 co-injection techniques. The barrier could alternatively
- 21 be acrylonitrile-methyl acrylate copolymer, ethylene
- 22 vinyl alcohol (EVOH) or nylon MXD6. The barrier could be
- 23 provided on the inside or outside of the container.

24

- 25 Oxygen scavengers such as all polyester Amosorb 3000 or
- 26 X-312 scavenger may be used. These Oxygen scavenging
- 27 materials can be incorporated into the material of the
- 28 container to react with the gas before it reaches the
- 29 contents. Amosorb 3000 or X-312 scavenger have particular
- 30 application when the barrier selected is MXD6 nylon. With
- 31 these types of active oxygen scavenging packages, shelf
- 32 life performance is determined solely by the rate of
- 33 carbonation loss and CO2 loss is reduced by the presence

. .:

I of the MXD6 as a physical barrier. A scavenger of MXD6

2 with metal catalysed oxygen reduction chemistry may also

3 be used (Oxbar). This system reacts very quickly with

4 oxygen in the container and has a high oxygen capacity,

5 ensuring a long active life.

6

7 The container may also have an inorganic coating such as

8 amorphous carbon. This can be sprayed onto the surface

9 of the container being coated. The inorganic coating can

10 be applied either to the inside or outside of the bottle

11 after blowing. Plasma-applied coatings, using carbon or

12 silicon, which have recently been developed, may be used.

13 The Sidel Actism and Kirin DLCm coating technologies can

14 be used produce a thin layer of amorphous carbon,

15 typically 100 to 200nm thick, on the inside surface of

16 the container. This is deposited from a high-energy

17 plasma of acetylene gas within a high vacuum environment.

18 The coating provides an excellent barrier to both O_2 and

19 CO2, and, because it is on the inside of the container,

20 prevents the O2 dissolved in the material of the container

21 from migrating into the contents of the container during

22 the first few weeks of storage.

23

24 Because the deposited layers are fundamentally brittle,

25 they have to be extremely thin in order not to flake off

26 under container stresses, caused by bottle expansion and

27 creep when the bottle is filled, and under pressure from

28 the contents. Other factors include damage and scuffing

29 due to bottle handling, but these clearly do not affect

30 the integrity of the coating if it is on the inside. The

31 barrier performance improvements of carbon coatings are

32 similar to those achieved by organic coatings, again

1 giving a longer potential retail shelf life of around.

2 nine months.

3

4 Silica technologies such as Glaskin and BestPet can also

- 5 be used. These rely on the application of a SiOx vacuum
- 6 plasma coating, to give a barrier layer between 40 and
- 7 60nm thick. While the Glaskin process applies the glass
- 8 clear coating to the inside of the bottle, the BestPet
- 9 technique applies it to the outside.

10

2.1

- 11 As an alternative an organic coating may be used.
- 12 External organic coatings have been known and used in the
- 13 industry since the early 1980s. In the mid 1990s, barrier
 - 14 coating solutions based on two component epoxyamine
 - 15 chemistry (Bairocade) were developed, first to lengthen
 - 16 the shelf life of the smaller soft drink sizes in hotter
 - 17 climates, and then for beer. These provide a
 - 18 transparent, glossy, external spray coating which is an
 - 19 excellent barrier to migration of CO2 and O2, and is
 - 20 unaffected by humidity. The low temperature thermoset
 - 21 cure provides a tough film, robust to filling and
 - 22 handling conditions.

23

- 24 Typically the coating will be applied to the container at
- 25 thicknesses between 6µm and 10µm, and allow the use of
- 26 standard resins and preforms with existing injection and
- 27 blow moulding equipment. The use of such coatings
- 28 provides a performance improvement which is around 19
- 29 times better than an uncoated container and translates
- 30 into a longer retail shelf life. The external organic
- 31 coating may be PVDC two component epoxyamine.

1 The alternative approach to improving the gas permeation

2 properties of the container material is to manufacture it

3 from multiple layers of the material. In other words,

two or more layers of the container may be combined to

5 act as an improved oxygen barrier. Final shape blowing

6 produces a bottle with up to seven different polymer

7 layers, which either act as a physical barrier to gas

8 permeation, or are chemically active in scavenging oxygen

9 from the material of the container and intercepting

10 oxygen diffusing in from outside.

- 11

14 **3** . . .

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12 The material herein described has an elevated glass

13 transition temperature, which is much higher than the

14 glass transition temperature of, for example, PET. PET

15 has a glass transition temperature that is lower than the

16 pasteurisation temperature used in the beer industry. As

17 a result when PET is used in the manufacture of bottles,

18 creep may occur during filling. In other words the

19 material expands, which causes deformity of the bottle.

20 This problem is eliminated using the material herein

21 described as the glass transition temperature is above

22 the pasteurisation temperature used during filling.

23

24 Furthermore, bottles made from PET are generally filled

25 using flash pasteurisation, as opposed to full

26 pasteurisation, which the industry prefers. Full

27 pasteurisation is generally more efficient which results

28 in a longer shelf life for the product. However full

29 pasteurisation is not generally used with PET materials.

30 A particular advantage of the material herein described

31 is that because it has an elevated glass transition

32 temperature, it can withstand full pasteurisation.

1 It has been discovered that using the above described

2 material a container such as a bottle, glass or tumbler

3 can be manufactured which does not cut, puncture or

4 otherwise damage human skin or tissue when broken. In

5 other words, the container will shatter into harmless

6 fragments, shards or pieces when broken.

7

8 A particular advantage of the container described herein,

9 lies in the fact that even though it does not shatter

10 into dangerous fragment when broken, it has a similar

11 quality feel as glass, and has improved aesthetic

12 qualities over existing plastics such as PET. The

%3' material herein described for use in manufacturing a

: 14 container, is relatively light and glass-like to touch

(34 15 and as it is a polymer is can be processed, for example

16 by including oxygen barriers during production. 3: 300

Tiportantly, the material is thicker than an equivalent

18 PET bottle so has a more glass-like feel but can be

19 manufactured into containers without an increase in cost.

20

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21 Example 1

22

- 23 An 85% mix of polystyrene polymer and 15% resin is used
- 24 to manufacture a bottle with improved safety
- 25 characteristics. The 15% resin maybe comprised of a
- 26 single resin selected from the group consisting of
- 27 Norsolene TM, Krystalex TM, Plastolyn TM, Endex TM, Sokorez
- 28 [™], Arkon [™], Piccolastic[™] and Piccotex[™], or may be a
- 29 combination of two or more of the above. Plastolyn TM is
- 30 particularly suitable for this purpose. The resin or
- 31 resins are selected to achieve a desired molecular weight
- 32 range.

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- 1 Further modifications and improvements may be added
- 2 without departing from the scope of the invention herein
- 3 intended.

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